

The C-2 Selective Azide Opening Reaction of *trans*-2,3-Epoxy Alcohols by NaN₃ and PhB(OH)₂

Hiroyuki Hayakawa, Naoko Okada, Masahiro Miyazawa, and Masaaki Miyashita*

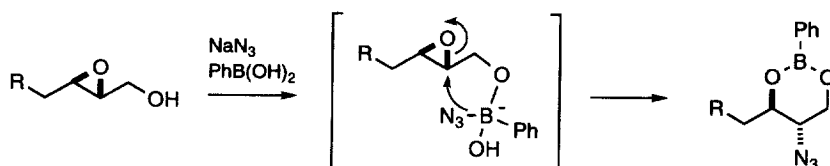
Division of Chemistry, Graduate School of Science, Hokkaido University, Sapporo 060-0810, Japan

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Abstract: The azide opening reaction of *trans*-2,3-epoxy alcohols with NaN₃ and PhB(OH)₂ in DMF occurred regio- and stereoselectively at the C-2 position giving rise to phenylboronates of *trans*-2-azido-1,3-diols in high yields. © 1999 Elsevier Science Ltd. All rights reserved.

Stereoselective epoxide-opening reactions have been recognized as one of the most important transformations in organic synthesis.¹ Among them, the azide opening reaction of 2,3-epoxy alcohols has gained importance in context with the synthesis of biologically active molecules.² Although various methods / reagents such as Me₃SiN₃-Et₂AlF,³ NaN₃-NH₄Cl,⁴ NaN₃ supported on calcium zeolite,⁵ [Ti(O-*i*-Pr)₂(N₃)₂],⁶ Bu₃SnN₃,⁷ Ti(O-*i*-Pr)₄-Me₃SiN₃,⁸ Et₂AlN₃,⁹ etc. have been developed for the purpose, the regioselectivity in these azide opening reactions is restricted only at the C-3 position. On the other hand, examples pertaining to the C-2 selective azide opening of 2,3-epoxy alcohols are few, though isolated cases arising out of steric requirements⁴ or the use of 2-trimethylsilyl-2,3-epoxy alcohols¹⁰ have been reported. We have now developed a methodology to achieve the C-2 selective azide opening of *trans*-2,3-epoxy alcohols by the combination of NaN₃ and phenylboronic acid (PhB(OH)₂).

Although the C-2 selective azide opening reaction of 2,3-epoxy alcohols devoid of any steric and/or electronic bias has never been reported, we anticipated that the ring opening of such epoxy alcohols with NaN₃ and PhB(OH)₂ may occur regioselectively at the C-2 position by directing effect of PhB(OH)₂, as shown in Scheme 1. Indeed, this was realized in practice very efficiently by using 3 equiv. of NaN₃ and 2 equiv. of PhB(OH)₂ in hot DMF to provide phenylboronates of the corresponding 2-azido-1,3-diols in high yields.



Scheme 1

The representative results are summarized in Table 1. As shown, the ring-opening reactions of *trans*-4-benzyloxy- and *trans*-5-benzyloxy-2,3-epoxy alcohols with NaN_3 and PhB(OH)_2 occurred exclusively at the C-2 position to give phenylboronates of *trans*-2-azido-1,3-diols as a single product in 99% and 94% yields, respectively (entries 1 and 2). Similarly, the reaction of an analogous substrate having a silyloxy group proceeded cleanly to afford the C-2 azido-compound predominantly (entry 3). Thus, all the reactions of disubstituted *trans*-2,3-epoxy alcohols with NaN_3 and PhB(OH)_2 provided the corresponding phenylboronates of *trans*-2-azido-1,3-diols exclusively or predominantly (entries 1-5). The stereochemistry of the derived products were confirmed unambiguously by NMR analyses and sodium periodate oxidation of the corresponding diols.¹¹ It is noteworthy that the phenylboronates thus obtained were very stable and could be easily purified by silica gel column chromatography and stored in a refrigerator for a long time without any appreciable decomposition.¹²

In contrast to the reactions of *trans*-2,3-epoxy alcohols, the corresponding *cis*-analogs were found to react sluggishly to provide lower yields of products presumably due to steric hindrance, though the reaction furnished the C-2 azide as the major product (entry 7). The result in Table 1 also demonstrates that the substrates having an ether oxygen in the side chain react faster and with higher regioselectivity than those having no such functionality (entries 1 and 2 vs. entries 4 and 5), which can be accounted for the chelation effect. It is also evident that the regioselectivity was not observed in the reaction of a trisubstituted *trans*-2,3-epoxy alcohol (entry 6).

It is of interest to note that the present procedure is the first example in realizing the C-2 selective azide opening reaction of disubstituted *trans*-2,3-epoxy alcohols having no particular steric and electronic bias. It may be noted that the reactions of *trans*-4-benzyloxy-2,3-epoxy alcohol (the substrate in entry 1) with $[\text{Ti(O-}i\text{-Pr)}_2(\text{N}_3)_2]$ in benzene (70 °C) and with $\text{NaN}_3\text{-NH}_4\text{Cl}$ in aqueous 2-methoxyethanol (124 °C) have been studied by Sharpless et al. to give a 6:1 and 1:1 mixture of the corresponding C-3 and C-2 azido-isomers, respectively.⁶ They have also observed that the reactions of *trans*-2,3-epoxy-1-hexanol (the substrate in entry 4) with $[\text{Ti(O-}i\text{-Pr)}_2(\text{N}_3)_2]$ in benzene (70 °C) and with $\text{NaN}_3\text{-NH}_4\text{Cl}$ in aqueous MeOH (65 °C) gave a 5.8:1 and 36:1 mixture of the corresponding C-3 and C-2 azides, respectively.⁶ Thus, the former methods/reagents have been known to produce 3-azido-1,2-diols regioselectively or predominantly.

In conclusion, we have developed a highly C-2 selective azide opening reaction of *trans*-2,3-epoxy alcohols through a combination of NaN_3 and PhB(OH)_2 . Since a wide variety of optically active 2,3-epoxy alcohols are readily available via the Katsuki-Sharpless asymmetric epoxidation,¹³ the present method will provide a convenient and selective way to prepare the phenylboronates of *trans*-2-azido-1,3-diols, thereby making it very useful in organic synthesis.

Table 1. Reaction of 2,3-Epoxy Alcohols with NaN_3 and PhB(OH)_2 in DMF.^{a)}

Entry	Substrate	Time (h)	Product Isolated	Yield
1		4.5		99%
2		18		94%
3		7		77%
				13%
4		27		78%
				22%
5		48 ^{c)}		81%
				14%
6		44		86% (5 : 7) ^{d)}
7		65		36%
				17%

^{a)} The reaction was carried out using NaN_3 (3 equiv.) and PhB(OH)_2 (2 equiv.) in DMF at 70°C.

^{b)} A chiral substrate.

^{c)} The reaction was carried out using NaN_3 (2 equiv.) and PhB(OH)_2 (1.3 equiv.) in DMF at 70°C.

^{d)} An inseparable mixture.

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- 11 The stereochemistry of two phenylboronates in entry 7 was confirmed by identification with that of authentic samples which were synthesized by phenylboronation of the corresponding 2-azido-1,3-diol and 3-azido-1,2-diol, respectively, readily prepared according to ref. 6.
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